

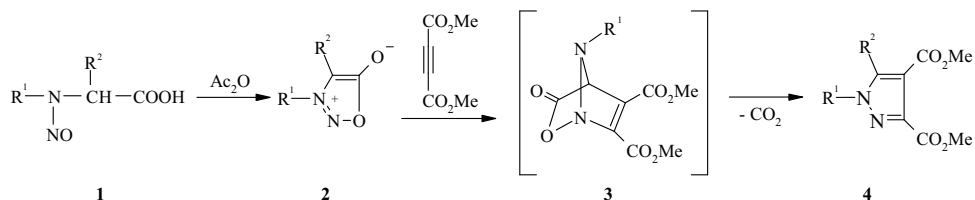
NEW HALOGENATED SYDNONES AND THEIR CYCLOADDITION TO FORM PYRAZOLES

F. Dumitrașcu*, Carmen Irena Mitan, C. Drăghici,
Loredana Barbu and M.T. Căproiu

New halogenated pyrazoles **9b-d** were obtained by 1,3-dipolar cycloaddition reaction of several halogenated sydnones **8b-d** with dimethyl acetylenedicarboxylate. The hydrogen bonds formation between 4-H proton of sydnones **8a,b** was put in evidence by ¹H-NMR. The structure of new compounds was assigned by elemental analysis, ¹H- and ¹³C-NMR spectra.

Introduction

Among the mesoionic compounds syndrones **2** are the best studied and thoroughly known [1÷5]. Sydnones can readily prepared by cyclodehydration of N-substituted-N-nitroso-aminoacids **1** with reagents such as acetic anhydride. The resulting compounds contain a mesoionic aromatic system, which can be depicted with polar resonance structures. Sydnones undergo smooth cycloaddition with acetylenes to give pyrazoles **4** in high yield [6÷9]. The reaction involves a 1,3-dipolar cycloaddition of the sydnones, behaving like a cyclic azomethine imine, to the corresponding acetylene followed by carbon dioxide evolution and aromatization (Scheme 1).



Scheme 1.

The present work describes the synthesis of new halogenated sydnones and their cycloaddition reaction to form pyrazoles. The halogen atoms are present in the benzene and/or heterocycle ring. The influence of steric effects on reactivity of sydnones is also discussed.

Experimental

All melting points were recorded with a Boetius microapparatus and are uncorrected. NMR spectra were recorded with a Varian Gemini 300 instrument, the chemical shifts being expressed in δ values relative to TMS as internal standard.

* Center of Organic Chemistry "C.D. Nenitescu", Spl. Independentei 202B, Sect.6, Bucharest

***N*-(4-Bromo-2-chlorophenyl)glycine (6b)**

A solution of 4.8 g (30 mmol) of bromine in 15 ml of glacial acetic acid was dropped under stirring to a suspension of 7.9 g (30 mmol) of *N*-(2-chlorophenyl)glycine (**5**) [10] in 25 ml of glacial acetic acid. Stirring was continued for 30 min. The reaction mixture was poured into water and the precipitate was filtered by suction. Mp 149–151°C (Lit. [11] 156–157 °C); yield 87%.

¹H-NMR (CDCl₃+TFA, δ_{ppm}, J_{Hz}): 4.52 (s, 2H, CH₂); 7.64 (dd, 1H, 8.6, 2.1, H-5'); 7.79 (d, 1H, 8.6, H-6'); 7.80 (d, 1H, 2.1, H-3').

¹³C-NMR (CDCl₃+TFA, δ_{ppm}): 51.1 (CH₂); 126.2 (C-4'); 126.5 (C-6'); 128.7 (C-2'); 129.8 (C-1'); 132.6 (C-5'); 134.3 (C-3'); 169.7 (CO).

Sydnonones 8a,b – General procedure

To a solution of 2 g NaOH in 30 ml of water were added 20 mmol *N*-arylglycine **6a,b** and 1.66 g (21 mmol) of NaNO₂. In the cooled solution 10 ml of HCl were dropped under stirring, the temperature maintained under 5 °C. The nitroso derivatives **7a,b**, which separated as oils, were extracted twice with CH₂Cl₂. The organic layer was dried on CaCl₂ and then the solvent was evaporated off. The residue was treated with 30 ml of acetic anhydride and 2 ml of pyridine and evaporated under reduced pressure on the water bath. The crude products **8a,b** were recrystallized from ethanol.

***3*-(2-Chlorophenyl)sydnone (8a) Mp 84-85°C (Lit.[12] 82–84°C)**

¹H-NMR (CDCl₃, δ_{ppm}): 6.65 (s, 1H, H-4); 7.60 (m, 4H, Ar-H).

¹H-NMR (DMSO-d₆, δ_{ppm}, J_{Hz}): 7.64 (s, 1H, H-4); 7.68 (td, 1H, 7.8, 1.7, H-5'); 7.79 (td, 1H, 7.8, 1.7, H-4'); 7.87 (dd, 1H, 7.8, 1.6, H-3'); 7.96 (d, 1H, 7.8, 1.6, H-6').

¹³C-NMR (CDCl₃, δ_{ppm}): 97.9 (C-4); 127.1 (C-6'); 128.3 (C-2'); 129.1 (C-5'); 131.4 (C-3'); 132.8 (C-1'); 133.4 (C-4'); 168.0 (CO).

¹³C-NMR (DMSO-d₆, δ_{ppm}): 99.9 (C-4); 128.9 (C-6'); 129.0 (C-2'); 129.6 (C-5'); 131.8 (C-3'); 132.9 (C-1'); 134.8 (C-4'); 169.0 (CO).

***3*-(4-Bromo-2-chlorophenyl)sydnone (8b) Mp 112-113°C (EtOH); yield 68%.**

¹H-NMR (CDCl₃, δ_{ppm}, J_{Hz}): 6.64 (s, 1H, H-4); 7.50 (d, 1H, 8.5, H-6'); 7.68 (dd, 1H, 8.5, 2.0, H-5'); 7.83 (d, 1H, 2.0, H-3').

¹H-NMR (DMSO-d₆, δ_{ppm}): 7.68 (s, 1H, H-4).

¹³C-NMR (CDCl₃, δ_{ppm}): 97.8 (C-4); 127.2 (C-4'); 128.6 (C-6'); 130.1 (C-2'); 131.4 (C-1'); 131.7 (C-5'); 134.1 (C-3'); 168.3 (CO).

Anal. Calcd. for C₈H₄BrClN₂O₂: N, 10.17; Found: N, 10.39 .

4-Iodosydnonones 8c,d – General procedure

A solution of 22 mmol (1.1 ml) of iodine monochloride in 10 ml of glacial acetic acid was added dropwise to stirred mixture of 20 mmol of sydnone **8a,b** and 2.2 g of (25 mmol) dry sodium acetate and of 15 ml glacial acetic acid. Stirring was continued for 2 hrs. at

55–60 °C, after which the 4-iodosydnone was precipitated by the addition of water. The product was filtered off and thoroughly washed with water.

3-(2-Chlorophenyl)-4-iodosydnone (8c) Mp 156-158°C (EtOH); yield 82%.

¹H-NMR (CDCl₃, δ_{ppm}): 7.50 (m, H-3'); 7.60 (m, H-5'); 7.71 (m, H,-4', H-6').

¹³C-NMR (CDCl₃, δ_{ppm}): 53.1 (C-4); 127.9 (C-6'); 128.3 (C-5'); 130.6 (C-2'); 131.2 (C-1'); 132.8 (C-1'); 133.8 (C-4'); 168.4 (CO).

Anal. Calcd. for C₈H₄ClIN₂O₂: N, 8.69; Found: N, 8.82.

3-(4-Bromo-2-chlorophenyl)-4-iodosydnone (8d) Mp 178-180°C (EtOH); yield 83%.

¹H-NMR (CDCl₃, δ_{ppm}, J_{HZ}): 7.38 (d, 1H, 8.5, H-6'); 7.73 (dd, 1H, 8.5, 2.0, H-5'); 7.88 (d, 1H, 2.0, H-3').

¹³C-NMR (CDCl₃, δ_{ppm}): 53.2 (C-4); 127.7 (C-4'); 128.9 (C-6'); 130.2 (C-2'); 131.7 (C-1'); 131.8 (C-5'); 134.0 (C-3'); 168.2 (CO).

Anal. Calcd. for C₈H₃BrClIN₂O₂: N, 6.98; Found: N, 7.26.

Pyrazoles 9b-d – General procedure

A mixture of 10 mmol of the sydnones **8b-d** and 1.7 g (12 mmol) of DMAD was refluxed in 30 ml toluene (4hrs. for **8b**) or 30 ml xylene (8hrs. for **8c,d**). After removal of the solvent in vacuo, the pyrazoles were recrystallized from ethanol.

1-(4-Bromo-2-chlorophenyl)-3,4-dicarbomethoxy-pyrazole (9b) Mp 105-106°C (EtOH); yield 91%.

¹H-NMR (CDCl₃, δ_{ppm}, J_{HZ}): 3.89 and 3.99 (2s, 6H, OCH₃); 7.50 (d, 1H, 8.5, H-6'); 7.56 (dd, 1H, 8.5, 1.9, H-5'); 7.73 (d, 1H, 1.9, H-3'); 8.32 (s, 1H, H-5).

¹³C-NMR (CDCl₃, δ_{ppm}): 52.1 and 52.8 (OCH₃); 116.0 (C-4); 123.6 (C-4'); 128.9 (C-6'); 129.5 (C-2'); 131.2 (C-5'); 133.3 (C-3'); 135.7 (C-1'); 136.2 (C-5); 144.7 (C-3); 161.6 and 161.7 (CO).

Anal. Calcd. for C₁₃H₁₀BrClN₂O₄: N, 7.50; Found: N, 7.79.

3,4-Dicarbomethoxy-1-(2-chlorophenyl)-5-iodopyrazole (9c) Mp 93-94°C (EtOH); yield 78%.

¹H-NMR (CDCl₃, δ_{ppm}, J_{HZ}): 3.92 and 3.94 (2s, 6H, OCH₃); 7.39 (dd, 1H, 7.6, 2.2 H-6'); 7.43 (td, 1H, 7.6, 1.7, H-5'); 7.50 (td, 1H, 7.6, 2.2, H-4'); 7.57 (dd, 1H, 7.6, 1.7, H-3').

¹³C-NMR (CDCl₃, δ_{ppm}): 52.3 and 52.7 (OCH₃); 91.4 (C-5); 121.0 (C-4); 127.6 (C-5'); 129.9 (C-6'); 130.3 (C-4'); 131.9 (C-3'); 132.8 (C-2'); 137.0 (C-1'); 145.4 (C-3); 161.2 and 162.0 (CO).

Anal. Calcd. for C₁₃H₁₀ClIN₂O₄: N, 6.67; Found: N, 6.90.

1-(4-Bromo-2-chlorophenyl)-3,4-dicarbomethoxy-5-iodopyrazole (9d) Mp 92-93°C (EtOH); yield 87%.

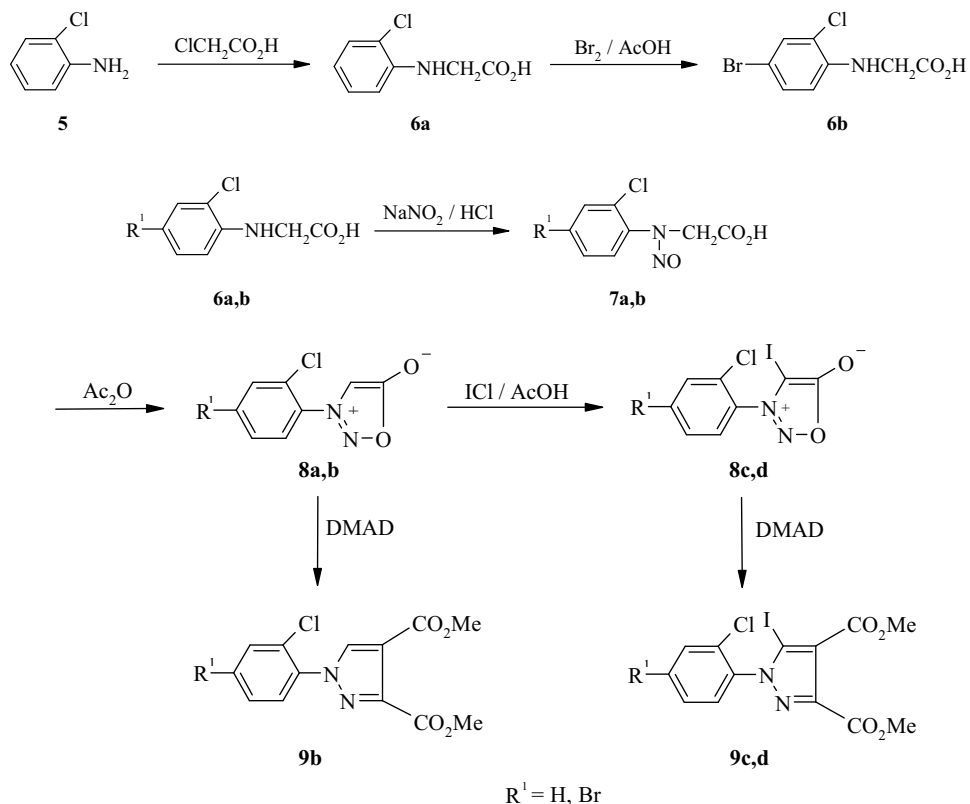
¹H-NMR (CDCl₃, δ_{ppm}, J_{HZ}): 3.94 and 3.96 (2s, 6H, OCH₃); 7.29 (d, 1H, 8.5, H-6'); 7.60 (dd, 1H, 8.5, 1.9, H-5'); 7.76 (d, 1H, 1.9, H-3').

^{13}C -NMR (CDCl_3 , δ_{ppm}): 52.3 and 52.8 (OCH_3); 91.4 (C-5); 121.2 (C-4); 125.3 (C-4'); 130.9 and 131.0 (C-5', C-6'); 133.1 (C-3'); 134.0 (C-2'); 136.0 (C-1'); 145.7 (C-3); 161.1 and 161.9 (CO).

Anal. Calcd. for $\text{C}_{13}\text{H}_9\text{BrClIN}_2\text{O}_4$: N, 5.61; Found: N, 5.90.

Results and Discussion

The sydrones **8a,b** were obtained from 2-chloroaniline according to Scheme 2. We prepared N-(4-bromo-2-chlorophenyl)glycine (**6b**) by bromination of **6a**. This represents a new method for preparation of **6b**. Recently [13] we obtained good results in the direct iodination of sydnone ring by using the reagent iodine monochloride/acetic acid. By using this method the sydrones **8a,b** could be iodinated with this reagent in the presence of an equivalent of sodium acetate added to neutralize the hydrochloride acid formed in the reaction. The products were obtained in a highly pure state and in yields of over 80%. Two new 4-iodosydrones **8c,d** were obtained by this method.



Scheme 2.

The chemical shift of the 4-H proton of sydrones **8a,b** in DMSO-d_6 appears as being unusually high ($\delta = 7.64$ and 7.68 ppm) as compared to those measured in other solvent as

CDCl_3 ($\delta = 6.65$ and 6.64 ppm). A plausible explanation is the formation of hydrogen bonds between DMSO and 4-CH group. This is supported by a ^{13}C -NMR study of sydnones, which confirm the tendency of 4-CH group to form hydrogen bonds [14].

The ^{13}C -NMR spectra of 4-iodosydnones showed a strong negative increment at C-4 ($\Delta\delta = 44.7$). A shielding effect of 3-aryl group on C-4 was also apparent, provided that the aromatic ring was not strongly deviated by coplanarity by ortho substituents.

The transformation of sydnones **8b-d** into halogenated 1-arylpyrazoles was performed by 1,3-dipolar cycloaddition reactions with dimethyl acetylenedicarboxylate (DMAD). The cycloaddition reaction of 4-iodosydnones **8c,d** required a higher boiling solvent (xylene) and a long period of time (8hrs.) than 4-unsubstituted sydnone **8b** (toluene, 4hrs.). The difference of reactivity is explained by steric effect of bulky 4-iodo substituent on the formation of intermediate cycloadduct of type **3**.

By this method three new halogenated pyrazoles **9b-d** were obtained. The ^{13}C -NMR spectra of 5-iodopyrazoles showed about the same negative increments ($\Delta\delta = 44.8$) for signal of C-5 as in the case of corresponding 4-iodosydnones ($\Delta\delta = 44.7$).

For 4-iodopyrazoles [15] negative increments of $\Delta\delta$ were measured.

All new halogenated compounds **8b-d** and **9b-d** were fully characterized by elemental analysis, ^1H -, ^{13}C - and NMR experiments, namely APT, HETCOR and HH-COSY.

Conclusion

Polihalogenated pyrazoles **9b-d** were obtained by 1,3-dipolar cycloaddition between 4-unsubstituted **8b** and 4-iodosydnones **8c,d** with dimethyl acetylenedicarboxylate.

The ^1H -NMR spectra in DMSO-d_6 of sydnones **8a,b** put in evidence the formation of the hydrogen bonds between 4-H atom of the sydnone ring and DMSO.

The structure of the new compounds **8b-d** and **9b-d** was assigned by elemental analysis and ^1H - and ^{13}C -NMR spectra.

REFERENCES

1. Baker W. and Ollis W.D. (1957) *Quart. Rev.* **11**, 15-29.
2. Stewart F.H.C. (1964) *Chem. Rev.* **64**, 129-147.
3. Ohta M. and Kato H. (1967) *Nonbenzenoid Aromatics*, 16-I, Academic Press Editor J.P. Snyder, New York, 117-170.
4. Ollis W.D. and Ramsden C.A. (1976) *Adv. Heterocycl. Chem.* **19**, 1-122.
5. Newton C.G. and Ramsden C.A. (1982) *Tetrahedron* **38**, 2965-3011.
6. Huisgen R. (1963) *Angew. Chem.* **75**, 604-637.
7. Huisgen R., Gotthard H. and Grashey R. (1968) *Chem. Ber.* **101**, 536- 551.
8. Gotthard H. and Reiter F. (1979) *Chem. Ber.* **112**, 1193-1205.
9. Meazza G., Zanardi G. and Piccardi P. (1993) *J. Heterocyclic Chem.* **30**, 365- 371.

10. Knoevenagel E. (1904) *Ber. Dtsch. Chem. Ges.* **37**, 4082-4086.
11. Takeda A. (1957) *J. Org. Chem.* **22**, 1096-1100.
12. ***Wiley P.F., *US Patent 3224937*, 1966, 64, 14194e.
13. Dumitrașcu F., Drăghici C., Dumitrescu D., Tarko L. and Răileanu D. (1997) *Liebigs Ann./Recueil* 2613-2616.
14. Hsien-Ju Tien S.M. and Cheng Kung S.Y.F. (1985) *Ta Hsueh Hsueh Pao, I Hsueh Pien* **20**, 97; Cf. *Chem. Abstr.*, 1987 107, 133782g.
15. Begtrup M., Boyer G., Cabildo P., Cativiela C., Claramunt R.M., Elguero J., Garcia J.I., Toiron C., Vedso P. (1993) *Magn. Reson. Chem.* **31(2)**, 107-168.